



# Alternative method for the measurement of formaldehyde and carbonyl compounds in indoor environments

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*Travail réalisé en collaboration avec l'Université de Pau et des Pays de l'Adour (Pierre Mocho, Laboratoire de Thermique, Energétique et Procédés)*

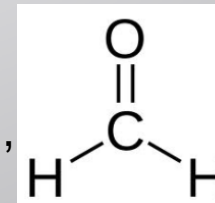


Paris, 21 juin 2011

# Formaldehyde and carbonyls compounds

## ■ Formaldehyde:

- Frequently found in indoor environments (building materials, combustion, chemical reactions...)
- « **Carcinogenic to humans** » (class 1, IARC), priority substance (OQAI)
- French Guide Value for Indoor Air: **10  $\mu\text{g.m}^{-3}$**  (long term exposure)



## ■ Acetaldehyde:

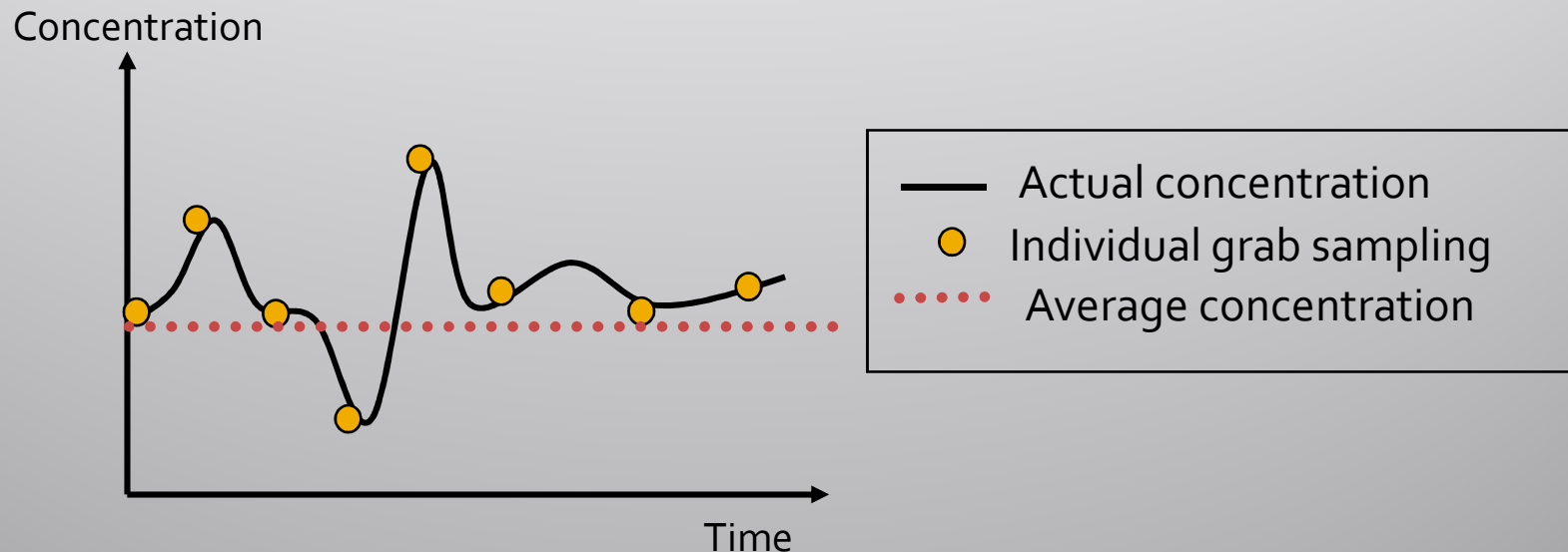
- Main sources: plywood products, chemical reactions,...
- « May be carcinogenic to humans » (class 2B, IARC), priority substance (OQAI)

## ■ Hexanal:

- Plywood panels, wood treatment products, solvent based paints, ...
- Relatively high concentrations: 1 – 106  $\mu\text{g.m}^{-3}$  (*dwelling campaign, OQAI, 2001*)

■ ...

# Two sampling strategies



## ➔ GRAB SAMPLING

Which gives information about the evolution of pollutant concentration.

## ➔ TWA SAMPLING

Which allows the determination of average concentration for the evaluation of long term exposure.

# Grab sampling: active method (NF ISO 16000-3, 2002)



On-site sampling with DNPH cartridges

Desorption in Acetonitrile



HPLC Analysis  
UV Detection



- ✓ Long sampling time ( up to 24 hours <sup>1</sup> for  $[\text{HCHO}]_{\text{air}} \approx 5 \mu\text{g} \cdot \text{m}^{-3}$ )
- ✓ Implementation: needs pump, flow-meter, tubing...
- ✓ Solvent desorption

<sup>1</sup> *Mesure des aldéhydes dans l'air ambiant à Paris*, Véronique Eudes et al., Laboratoire Central de la Préfecture de Police de Paris

# TWA sampling: passive method (NF ISO 16000-4, 2006)



Radiello® tubes



UME<sup>x</sup> Sampler

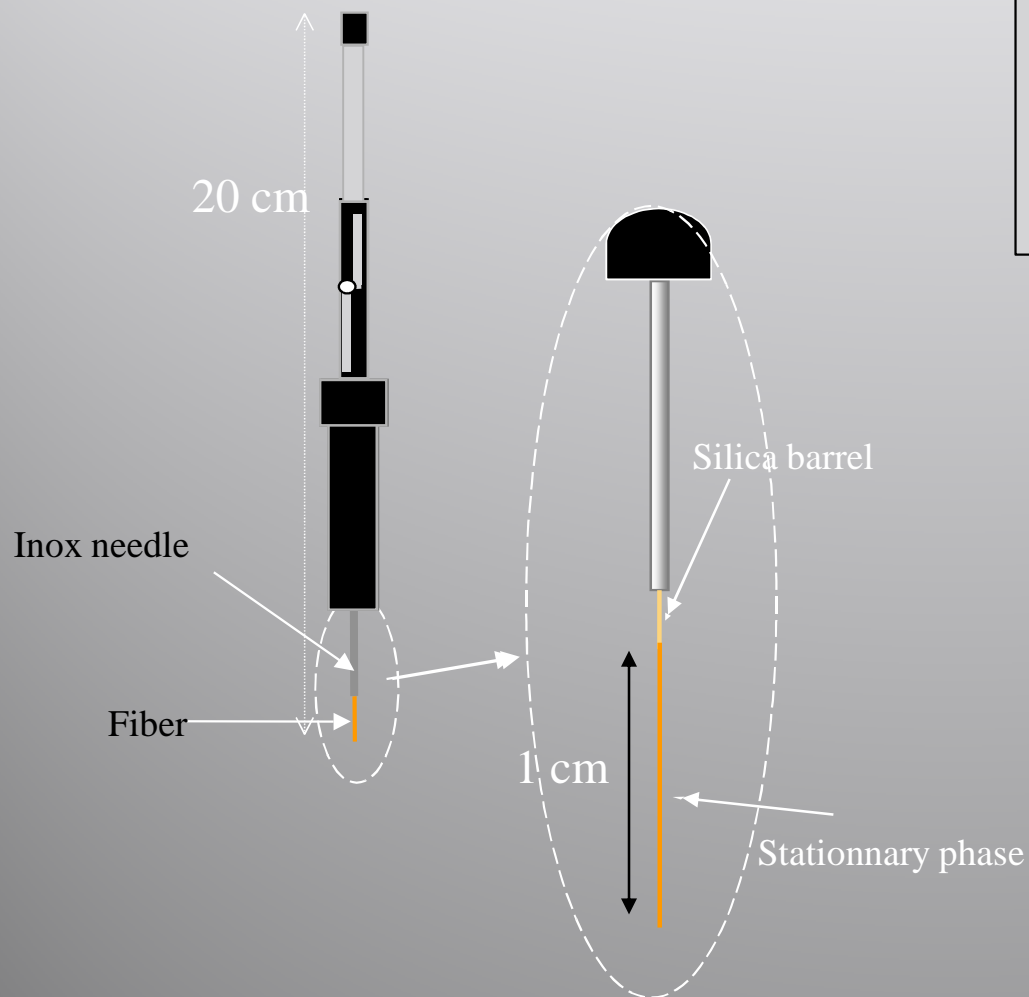


- ✓ No pump
- ✓ Easy to use
- ✓ Time weighted average

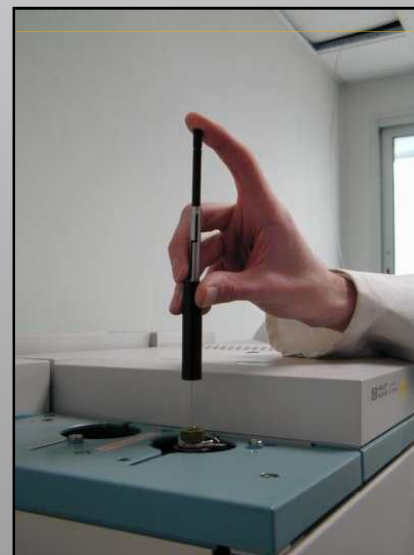


- ✓ Solvent desorption

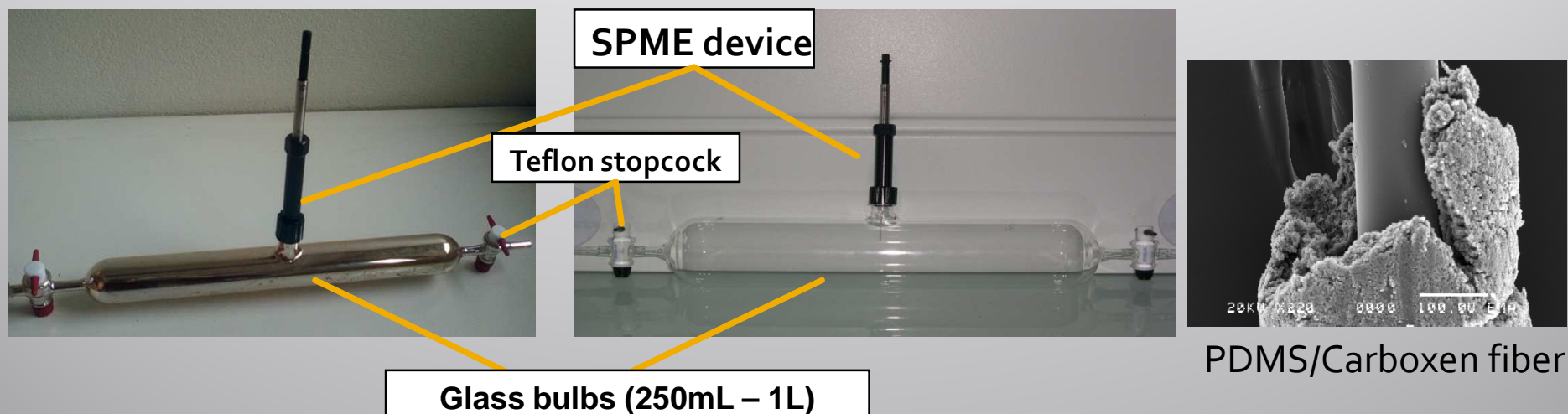
# Solid Phase Micro-Extraction (SPME): An Alternative Passive Technique



- Easy to use (passive sampling)
- One step thermal desorption in a classic GC injection port
- Solvent-free technique



# First application: Grab sampling of VOCs<sup>2</sup>



- Air sample is enclosed in a glass bulb
- A SPME fiber is exposed to the air: PDMS/Carboxen fiber chosen
- GC Analysis with FID or MS detection

Equilibrium:

3 hour extraction in 250 mL

→ Mean LOD:  $0.2 \mu\text{g} \cdot \text{m}^{-3}$

Non-equilibrium

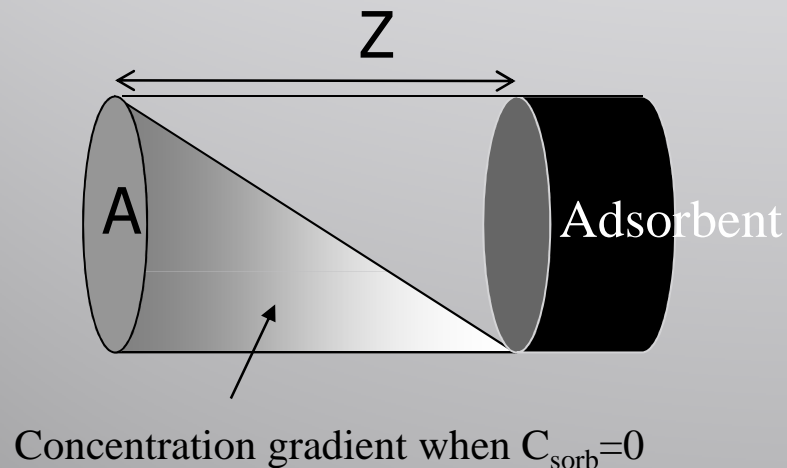
First Fick's Law

<sup>2</sup> Larroque V, Desauziers V, Mocho P, J. *Environ. Monit.*, 2006, **8**, 106-111



# First Fick's law of diffusion

➡ Used for passive samplers



$$n_f = D * \frac{A}{Z} \int (C_a - C_{sorb}) dt$$

$n_f$  : Extracted amount on the adsorbent ( $\mu\text{g}$ )

$D$  : Diffusion coefficient ( $\text{cm}^2.\text{min}^{-1}$ )

$A$  : adsorbent section ( $\text{cm}^2$ )

$Z$  : Diffusion length (cm)

$C_a$  : concentration in air ( $\mu\text{g}.\text{cm}^{-3}$ )

$C_{sorb}$  : Surface concentration ( $\mu\text{g}.\text{cm}^{-3}$ )

$t$  : Extraction time (min)

➡ At the beginning of sampling,  $C_{sorb}$  can be neglected:

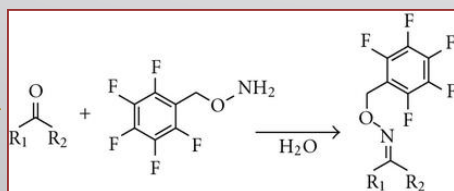
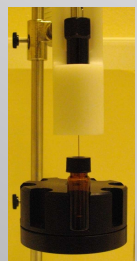
$$n_f = D * \underbrace{\frac{A}{Z}}_{\text{Uptake rate (U)}} * C_a * t \quad \longrightarrow \quad n_f = U. (C_a * t)$$



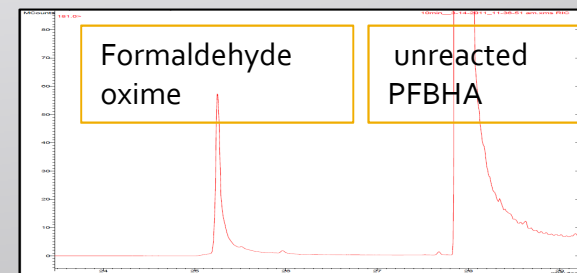
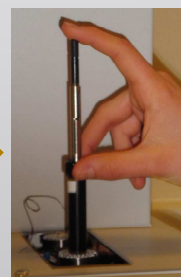
# Performances for VOCs analysis (GC/MS)

Compound	Ions for quantification (m/z)	Limits of detection in $\mu\text{g.m}^{-3}$ (1L – 4 min)	Limits of quantification in $\mu\text{g.m}^{-3}$ (1L - 4 min)	Reproducibility (n=5) (% RSD)
Acetone	43+58	1.4	4.5	5
Benzene	78	2.2	7.4	4
Cyclohexane	84+56	0.7	2.2	20
Trichloroethylene	95+130	0.8	2.5	9
Toluene	91	2.9	9.5	6
Butyl acetate	56+43	2.7	9.2	19
p-xylene	91+106	0.5	1.8	10
$\alpha$ -Pinene	93	0.4	1.3	7
n- decane	57+43	5.3	17.5	12
		Mean = 1.9	Mean: 6.2	Mean = 10.2

# Formaldehyde SPME sampling



Example: reaction  
with PFBHA



GC/MS analysis  
(PDMS-DVB fiber)

- Loading of the derivatization reagent on the fiber
- On-fiber derivatization during the exposure to gaseous formaldehyde <sup>3</sup>
- Thermal desorption into a GC injector
- GC analysis (FID or MS detection)

<sup>3</sup> Jacek A. Joziel, Japeth Noah and Janusz Pawliszyn, *Environ. Sci. Technol.* 2001. **35**, 1481-1846

# Development of the analytical method for formaldehyde SPME sampling

Three different coatings:

- PDMS/Carboxen
- PDMS/ DVB\*
- PDMS/DVB\*/Carboxen

\* DiVinylBenzene

Stationary Phase	BET** Surface (m <sup>2</sup> .g <sup>-1</sup> )	Porous volume (mL.g <sup>-1</sup> )		
		Micro	Meso	Macro
DVB	720	0.29	0.26	0.23
Carboxen	750	0.11	0.85	0.58

\*\* BET: Brunauer, Emmet, Teller

Two derivatization reagents:

- Pentafluorophenylhydrazine (PFPH)
- (o-2,3,4,5,6-pentafluorobenzyl)hydroxylamin (PFBHA)

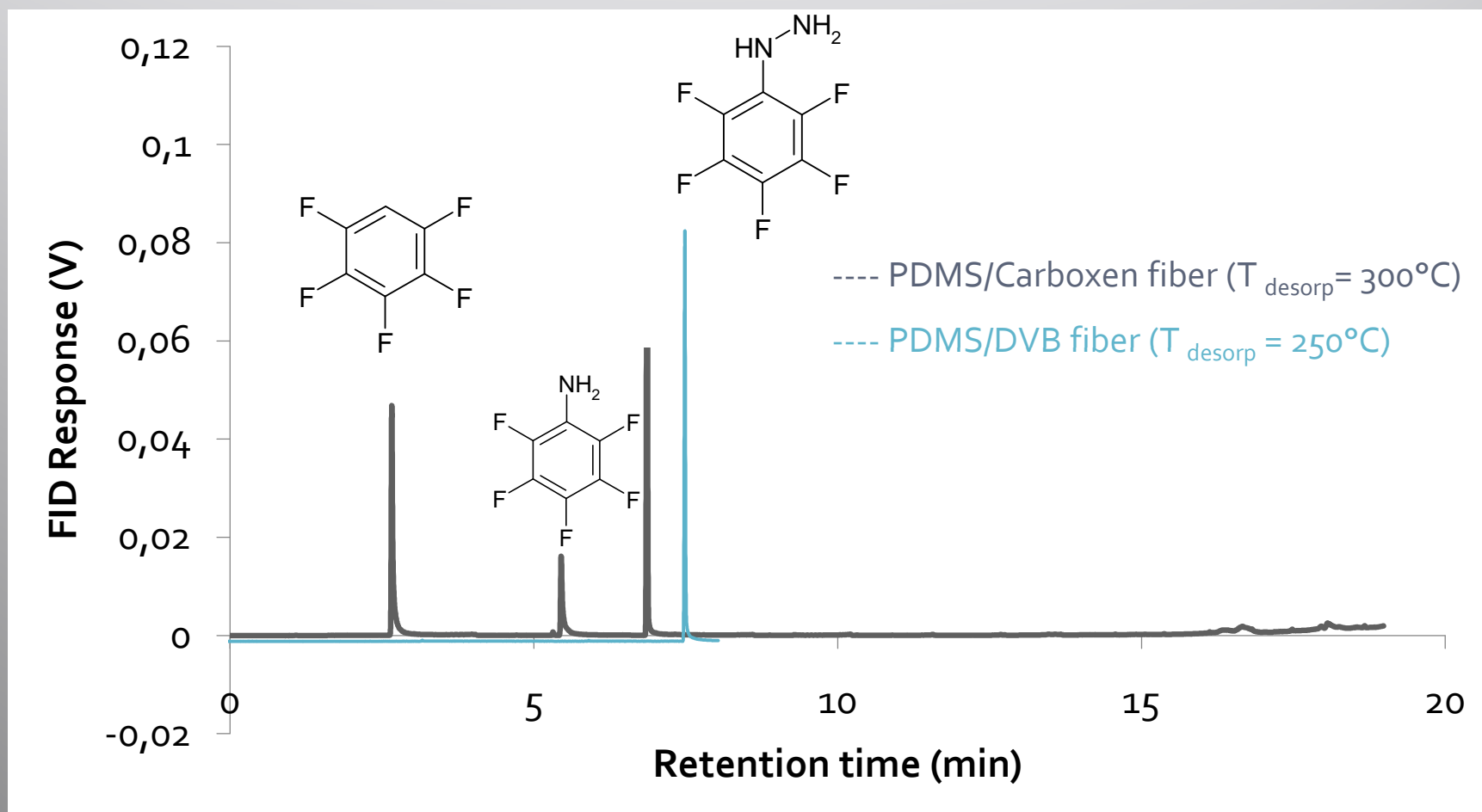


PDMS/DVB/Carboxen fiber

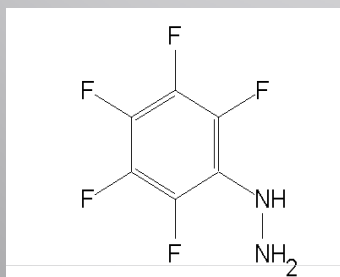
Loading processes:

- Pure product heated
- From an aqueous solution

# Choice of the fiber coating



# Choice of the derivatization reagent

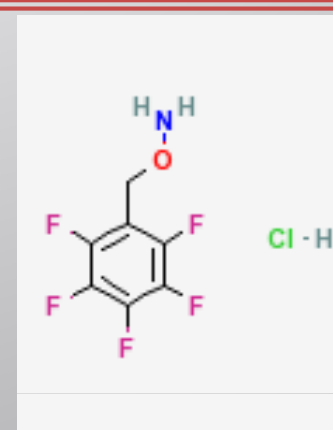


**PFPH**



Derivatization not easy to implement

- PDMS/DVB Fiber loaded with the derivatization reagent
- Fiber exposed to standard atmospheres of formaldehyde (permeation device)
- Analysis by GC/MS of the derivatives formed on the fiber



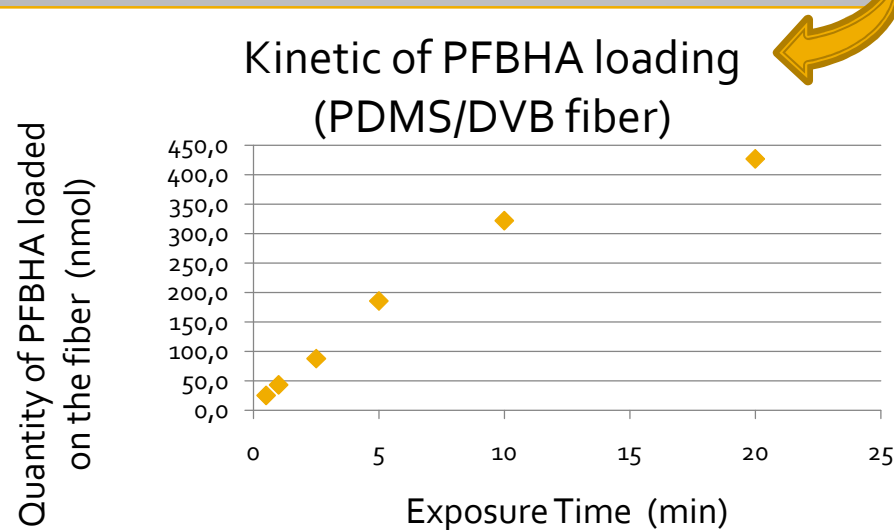
**PFBHA**



On-fiber derivatization easily performed at room temperature

# Development of the loading process

	Pure product 50 c	Aqueous solution 23 C
Equilibrium time	10 min (?)	20 min
Quantity of PFBHA loaded on the fiber (nmol)	80	420
Reproducibility (% RSD)	32	5



Loading applied:

**30 s**

↕

25 nmol PFBHA

↕

0.75  $\mu\text{g}$  HCHO...

↕

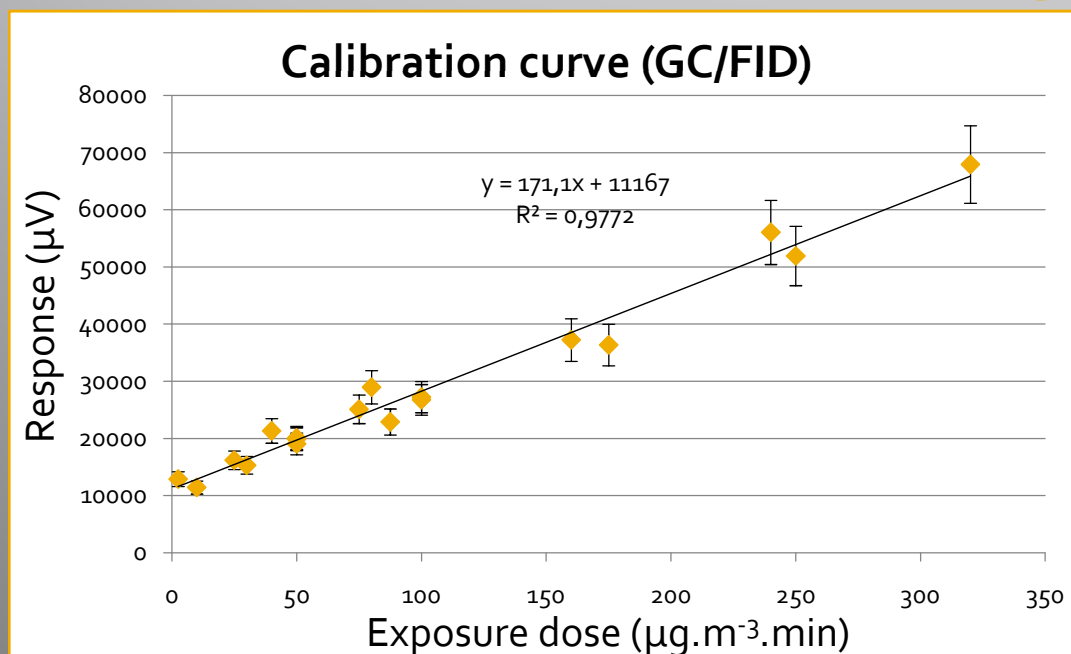
...in 250 mL: 3  $\text{mg}\cdot\text{m}^{-3}$

# Analytical Performances

- Fiber exposed to standard atmospheres
- SPME extraction in static mode into a 330 mL glass bulb

Formaldehyde concentration (C):  $2.5 - 80 \mu\text{g.m}^{-3}$   
 Exposure time (t):  $0.5 - 60 \text{ min}$

Exposure dose  $C \cdot t$  ( $\mu\text{g.m}^{-3}.\text{min}$ )



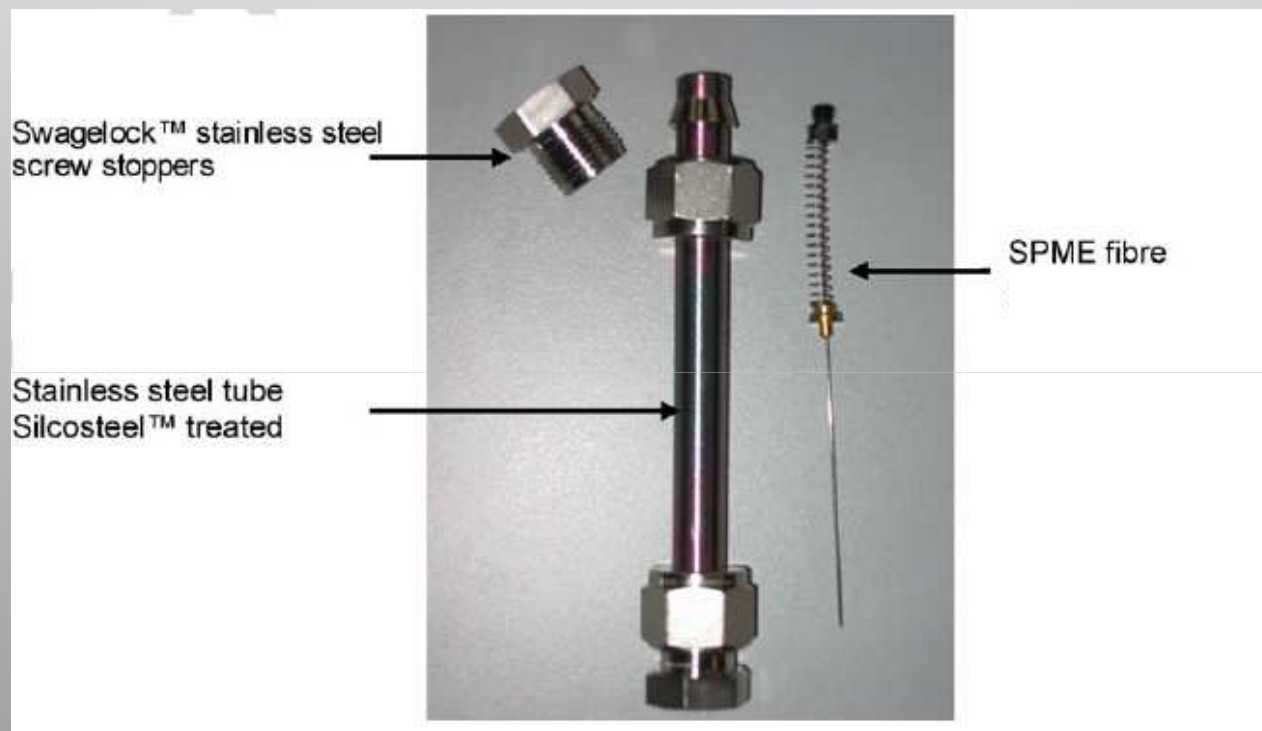
	GC/FID	GC/MS SIM Mode (m/z 181)
LOD	$0.9 \mu\text{g.m}^{-3}$	$1.2 \mu\text{g.m}^{-3}$
LOQ	$2.7 \mu\text{g.m}^{-3}$	$3.7 \mu\text{g.m}^{-3}$

20 min exposure

Repeatability: 14% RSD (n=6)



# Sample storage<sup>4</sup>



- 3 days before sampling: 77% (n=6)
- 3 days after sampling: 100 % (n=6)

<sup>4</sup> Virginie Larroque, Valérie Desauziers et Pierre Mocho, *J. Chromatogr. A*, **1124** (2006) 106-111

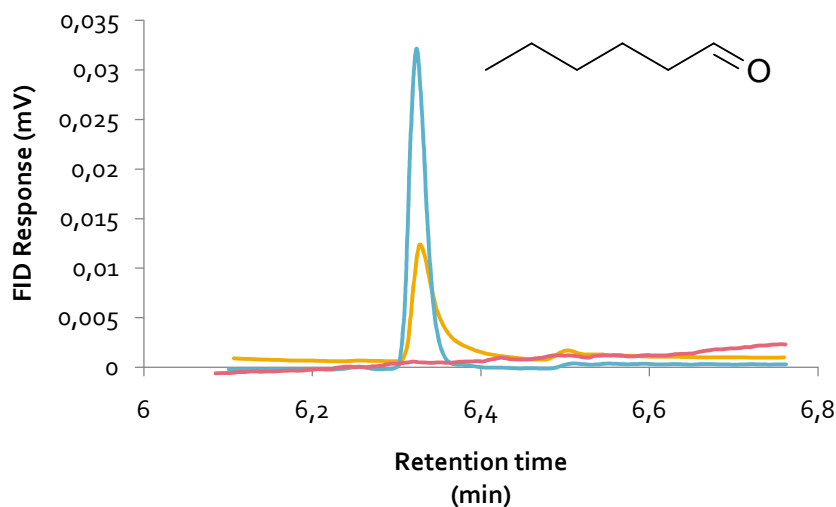
# Other carbonyl compounds: hexanal

[Hexanal]<sub>air</sub> = 510 µg.m<sup>-3</sup>

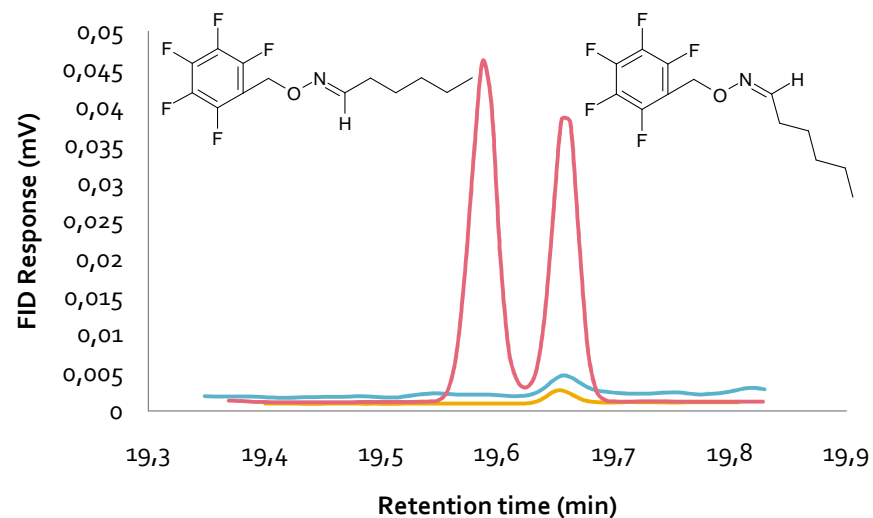
GC/FID

Exposure time: 15 min

## Hexanal



## Oximes PFBHA-Hexanal



----- PDMS/CAR ( $T_{\text{desorp}} = 300^{\circ}\text{C}$ )    - - - - - PDMS/DVB ( $T_{\text{desorp}} = 250^{\circ}\text{C}$ )    - - - - - PDMS/DVB + PFBHA ( $T_{\text{desorp}} = 250^{\circ}\text{C}$ )

- ✓ Quantitative reaction
- ✓ Higher sensitivity
- ✓ Apparition of two peaks instead of one => carbonyl compound

# Conclusion

## Analytical Development

### We have...

- ✓ Sensitive, fast and easy-to-use analytical method for carbonyl compounds

➡ LOD <  $1\mu\text{g.m}^{-3}$  with a 20 min sampling (formaldehyde)

### Next step....

- ✓ Validation: comparison with the standard method (NF ISO 16000-3)
- ✓ Combined sampling of VOCs and carbonyl compounds (2 fibers in one bulb)
- ✓ Passive sampler for on-site measurement of material emissions in buildings

# Future Prospects

**Application of these methods in new buildings**

➤ Kinetic study of building material emissions



**Development of a prediction model for indoor air quality**



**Health recommendations for the delivery of new or renovated buildings**

**Help for the new french regulation**

Thank you for your attention.

Questions?

